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4. The Normal Freezing Point Of Ethanol Was Obtained From The CRC Handbook Of Chemistry & Physics [3] And Found To Be -114.1°C At 0.1 MPa (roughly Atmospheric Pressure). The Freezing Point Depression Of The Solution Of Methanol In Ethanol Was Then Calculated Using This Normal Freezing Point Of Ethanol And Equation (17). Mar 12th, 2024

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Bonesteel, And E. M. Pierce, Academic Press, New York,1998. (We Will Give Detailed Procedure For Each Planned Experiment; However, This Lab Manual Can Help You Understand Much Better On Other Polymerizations, And On Characterization Techniques. In Addition, It Offers Detailed Information On The Experiments Covered In This Course) 2 • Lab Notebook With Pre-numbered, Carbonless Duplicate Pages ... Feb 24th, 2024

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