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[Type Text] [Type Text] [Type Text] LESSON 1 SOME BASIC CONCEPT OF CHEMISTRY Chemistry 1) 40 ml of Methane is completely burnt using 80 ml of Oxygen at room temperature. The volume of gas left after cooling to room temperature is (a) 40 ml CO<sub>2</sub> gas (b) 40 ml CO<sub>2</sub> gas and 80 ml H<sub>2</sub>O gas (c) 60 ml CO<sub>2</sub> gas and 60 ml H<sub>2</sub>O gas (d) 120 ml CO<sub>2</sub> gas Jan 8th, 2024

### UNIT 1 SOME BASIC CONCEPTS OF CHEMISTRY Chemistry

2 Prefixes In SI System Multiple Prefix Symbol 10<sup>-12</sup> Pico P 10<sup>-9</sup> Nano N 10<sup>-6</sup> Micro  $\mu$  10<sup>-3</sup> Milli M 10<sup>-2</sup> Centi C 10<sup>-1</sup> Deci D 10 Deca Da 10<sup>2</sup> Hecto H 10<sup>3</sup> Kilo K 10<sup>6</sup> Mega M 10<sup>9</sup> Giga G 10<sup>12</sup> Tera T Mass And Weight-- Mass Of A Substance Is The Amount Of Matter Present In It While Weight Is The Force Exerted By Gravity On An Object. The Mass Of A Substance Is File Size: 430KB Feb 22th, 2024

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In Which Any Number Can Be Represented In The Form  $N \times 10^N$ . N (Where N Is An Exponent Having Positive Or Negative Values And N Can Vary Between 1 To 10). E.g. We Can Write 232.508 As 2.32508  $\times 10^2$ . In Scientific Notation. Similarly, 0.00016 Can Be Written As 1.6  $\times 10^{-4}$ . Precision Refers Jan 2th, 2024

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