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9.) Using The Attached Table, Estimate The Change In Energy For The Following Reaction.  $\text{CH}_2\text{CH}_2(\text{g}) + \text{Br}_2(\text{l}) \rightarrow 2 \text{BrCH}_2\text{Br}(\text{l})$   $-92 \text{ KJ/mol}$  10.) Given The Following Information Calculate The Lattice Enthalpy Of Calcium Chloride. Page 6 Of 7 Apr 5th, 2024

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