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12. Reaction Given Below Is An Example Of  $\text{NCl}_2 + \text{Cu} + \text{HCl} \rightarrow \text{N} + \text{CuCl}_2 + \text{Cl}_2$  +- (1) Sandmeyer Reaction (2) Gattermann Reaction (3) Swarts Reaction (4) Gattermann - Koch Reaction 13. Identify Product F In The Following Sequence Of Reaction  $\text{NCl}_2 + \text{HBF}_4 \rightarrow \text{A} \rightarrow \text{BC} \rightarrow \text{CH}_3\text{NaNO}_2 \rightarrow \text{Cu} \xrightarrow{\Delta} \text{Sn} + \text{HCl} \rightarrow \text{HO}_2\text{Na} \rightarrow \text{NaN}$  Apr 4th, 2024

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De Nition 1.2: Heterogeneous Mixture A Heterogeneous Mixture Is One That Is Non-uniform And The Di Erent Components Of The Mixture Can Be Seen. 1.2.2  
Homogeneous Mixtures A Homogeneous Mixture Has A De Nite Composition, And Speci C Properties. In A Homogeneous Mixture, The Di Erent Parts Cannot Be Seen.  
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