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$2\text{S(g)}$ ,  $\Delta H^{\circ} = -20.2 \text{ kJ/mol}$  And  $\Delta S^{\circ} = +43.1 \text{ J/K}\cdot\text{mol}$ . Which Of These Statements Is True? A. The Reaction Is Only Spontaneous At Low Temperatures. B. The Reaction Is Spontaneous At All Temperatures. C.  $\Delta G$  2th, 2024

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