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Interplay Of Electron-Electron And Electron-Phonon

...One Of The Main Characteristics Of MJs In

Comparison To Large QDs Is The Strong Coupling Of Electrons To The Vibrations Of Molecule. This Electron-phonon (e-ph) Interaction Is One Of The Most

Important Candidates To Theoretically Explain

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Electron Arrangement Of The Electron Configuration Of

Atoms The Distance From The Bottom Of The Chart

Indicates The Energy Of Each Energy Level And Sub-

level. The Closer The Energy Level Is To The Bottom Of The Chart, The Lower Its Energy. At The Bottom Of The

Chart, You Will Find The First Energy Level, $n = 1$. The

Chart Shows Only One Circle In The First Energy Level.

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Chemical Periodicity 8.1 Characteristics Of Many-

Electron Atoms 8.2 The Quantum-Mechanical Model

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