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Chapter 3 Chemical Reactions And Reaction

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Following Balanced Equation. Identify The Limiting Reactant When 1.150 Grams Of HgO React With 12.46 Grams Of Cl<sub>2</sub>. Convert To Mol Jan 4th, 2024.

Chemistry 11 Stoichiometry Review Worksheet Name: KEY

Stoichiometry Review Worksheet Name: \_\_\_KEY\_\_\_ 33.  $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actual yield = 1.45g Theoretical yield = 5.95g  $\text{KClO}_4 \times 1 \text{ mol } 122.55 \text{ g} \times 2 \times 2 = 1 \text{ KClO}_4 \times$

$32.00 \text{ g } 1 \text{ mol} = 3.11 \text{ g O}_2$  %Yield =  $\frac{1.45 \text{ g } 3.11 \text{ g}}{1.45 \text{ g}} \times 100\% = 46.6\%$  34.  $4\text{FeCO}_3 + \text{O}_2$

$\rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$  Percent purity = 62.8%? pure FeCO<sub>3</sub> Mar 15th, 2024

Chemistry 12, Stoichiometry Review Problems, R. Janssen ... In A Camping Stove? 2. How Many

Moles Of O<sub>2</sub> Molecules Should Be Supplied To Burn 1 Mol Of CH<sub>4</sub> ... Was At One

Time Considered For Use As A Rocket Fuel. ... How Many Formula Units Of Calcium

Oxide (CaO) Feb 12th, 2024 Review Stoichiometry Modern Chemistry Answers E3

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Of The McQuarrie Textbook. Note That Chem 11 Problems Will Be More Involved  
And More Rigorous Than These! An Answer Key Is Provided At The End Of This  
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