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May Be Determined From The Formula
 $(\text{mol/L}) = 12.2 \frac{P(\text{atm})}{T-1(\text{K})}$ Where P And T Are The
Absolute Pressure And Temperature Of The Gas. The

Following Data Are Taken For The Feed Gas: $T=750\text{F}$, $P=150\text{ Psig}$, $H=210\text{ Mm}$, $R=82.4$ And For The Outlet Gas, $R=11.6$. Determine The (a) Mass Fraction Of SO_2
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To The Waste Lagoon (kg/h) And The Mass Fraction Of Cr In The Liquid (kg Cr/kg). Ans. 5779.8 Kg/h, 0.0154 Kg Cr/kg 9. In The Production Of Bean Oil, Bea Apr 3th, 2024

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