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Reaction: $3A \rightarrow 2B$ The Average Rate Of Appearance Of B Is Given By $D[B]/Dt$.
Comparing The Rate Of Appearance Of B And The Rate Of 11th, 2024PRE-IELTS 1
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CHEMICAL KINETICS - VSSUT2and Cl 2to FormHCl. 7 Chemistry-I. Figure 1.1: Plot Of
Concentration Against T For A Reaction $H_2 + Cl_2 \rightarrow 2HCl$ The Rate Of This Reaction
Does Not Depend On Concentration. 1.6 First Order Kinetics A Reaction Of The first
Order Is Represented As $X \rightarrow Y$ Where X Is The Reactant And Y The Product. The Rate
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