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10 When Vanadium(II) Compounds Are Dissolved In

Water, The Following Equilibrium Is Established. $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$

What Would Alter The

Composition Of The Equilibrium Mixture In Favour Of

The V^{2+} Ions? A Adding An Acid B Adding A Reagent

That Selectively Precipitates May 3th, 2024

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(c) (i) Write A Chemical Equation Representing The
Lattice Energy Of AgBr. (ii) Use The Following Data To
Calculate A Value For The Lattice Energy Of AgBr(s).
First Ionisation Energy Of Silver = $+731\text{kJmol}^{-1}$ Electron
Affinity Of Bromine = 325kJmol^{-1} Enthalpy Change Of
Atomisation Of Silver = $+285\text{kJmol}^{-1}$ Enthalpy Change
Of Atomisation Of Bromine = $+112\text{kJmol}^{-1}$... Jun 3th,
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