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Mole Conversion Chart Grams → Moles → Particles 1.62 G CaCl₂ $\times 1 \text{ Mol} / 110.98 \text{ g}$
 $\times 6.022 \times 10^{23} \text{ Particles Mol}^{-1}$ Practice Problems: 1. A Sample Of Neon Has A
Volume Of 75.8 L At STP . Calculate The Number Of Moles. 3.38 Mol Ne 2. Calculate
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