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Relationship, So 0.0322 Mol Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play A Role, Since It Is In Excess. Then, $n(\text{N}) = 0.0322$. aniline $n(\text{Aceta II})$. Ide Mol We Find The Molar Mass Of Acetanilide: $(\text{N}) = 128 + 19 + 14 + 16 + 1 = 135$. 89 M Aceta ... 2th, 2024

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