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Ag 108 G Ag 1 Mol Cu 1 Mol Ag 63.5 G Cu  
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Problem Sheet 2 Directions: Solve Each  
Of The Following Problems. Show Your  
Work, Including Proper Units  
1.  $\_\_ \text{CaCl}_2 + \_\_ \text{AgNO}_3 \rightarrow \_\_ \text{Ca}(\text{NO}_3)_2 + \_\_ \text{AgCl}$   
How Many Grams Of Silver Chloride  
Are Produced When 45 G Of Calcium  
Chloride React With Excess Silver ...  
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Stoichiometry Volume Problems  
Worksheet Answers  
Stoichiometry: Mixed Problems  
(KEY) 1)  $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$   
What Volume Of  $\text{NH}_3$  At STP Is  
Produced If 25.0 Of  $\text{N}_2$  Is Reacted  
With An Excess Of  $\text{H}_2$ ? 3 3 3 2 3 2 2 2 40.0L

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Name: \_\_\_KEY\_\_\_ 33.  $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actualyield=1.45g Theoyield=5.95g  $\text{KClO}_4 \times 1\text{mol}$

$122.55\text{g} \times 2\text{O}_2 \times 32.00\text{g} \times 1\text{mol} = 3.11\text{gO}_2$

%Yield=  $1.45\text{g} \times 3.11\text{g} \times 100\% = 46.6\%$  34.  $4\text{FeCO}_3 + \text{O}_2$

$\rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percentpurity=62.8%?gpureFeCO 3 12th, 2024

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