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4. $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$
5. $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$
Classification Of Chemical Reactions Reactions, Much Like Elements And Compounds, Can Be Classified According To Type. The Ability To Recognize And Classify Reactions Can Help Us Predict Jun 1th, 2024
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Section One: Reactions ...Section One: Reactions And Equations Chemical Reaction:
The Process By Which The Atoms Of One Or More Substances Are Rearranged To
Form Different Substances Products: Right Side Of The Reaction Reactants: Left
Side Of The Reaction Chemical Equation: Represents With Symbols And Formulas
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