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Section 3 Classifying Chemical ReactionsDecomposition Decomposition Reaction- One Substance Breaks Down, Or Decomposes, Into Two Or More Substances. AB → A + B Most Decomposition Reactions Require The Use Of Heat, Light, Or Electricity. 7th, 2024Classifying Chemical ReactionsClassifying Chemical Reactions Types Of Chemical Reactions: 1.

Synthesis 2. Decomposition 3. Combustion 4. Single-Replacement 5. Double-replacement \*Classifying How Atoms Rearrange During A Chemical Reaction. December 12, 2014 1. Synthesis Reaction: 2 Or More Substances React To Produce A Single Product A + B AB 8th, 2024Classifying Chemical Reactions WorksheetClassifying Chemical Reactions Worksheet Name: Period: \_\_\_\_\_ Classify Each Reaction As Acid/base, Redox, Synthesis, Decomposition, Single Replacement, Double Replacement Or Combustion. They May Be More Than One. The Equations Are Not Balanced. 1. 3th, 2024. Section 8-3: Classifying Chemical Reactions Reaction Type General Format Characteristics Of Reactants Characteristics Of Products Example (s) Decomposition AB A + B Only One Reactant Two Or More Simpler Products CuCO 3 (s) CuO + CO 2 (g) 2 H 2 O (l) 2 H 2 (g) + O 2 (g) Synthesis AKA: Combination O (l) A + B AB 2

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