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Interacts With Solvent Molecules. The Tert-butoxide Ion Is Very Bulky, Or Sterically Hindered, And Is Less Capable Of Interacting With The Solvent. The Ethoxide Ion Is Not As Sterically Hindered So It Can Accommodate More Solvent Interactions. As A Result Ethoxide Is Better Solvated And Is Therefore More Stable Than Tert-butoxide (Figure 3.7). 1th, 2024

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C) 1,1-dibromo-2-chloro-2-fluoropropane D)1,1-dibromo-2,2 Dichlorofluoropropane
12 Ethyne Is A(n) A) Alkane B) Alkene C) Alcohol D) Aromatic E) Alkyne 13 How Many Alkyne Isomers Are Possible With The Molecular Formula C₆H₁₀? A) 3 Or Less B) 4 C) 5 D) 6 E) 7 Or More 14 Alkanes Have The General Formula A) C_nH_{2n+2} B) C_nH_{2n} C) C_nH_{2n-2} D) C_nH_n E ... 1th, 2024

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The Figures Represent The Same Two Chemicals In Two Different Ways. The Formula That Shows The Bonds As Lines And What Atom Is Connect To What Are Called "structural Formulas Because From The Formula You Can See How The Atom's Are Attached To Each Other. The Lower Formulas Are Called Condensed Formulas . 3th, 2024

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