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3)  $2(aq) + Y$  C.  $K(s) + H_2SO_4(aq) \rightarrow Y$  D.  $Zn(s) + NaOH(aq) \rightarrow Y$  SECTION 11.3 REACTIONS IN AQUEOUS SOLUTION 1. Write The Net Ionic Equation For The Reaction Between Aqueous Barium Nitrate,  $Ba(NO_3)_2$ , And Sodium Sulfate,  $Na_2SO_4$ . 2. Magnesium Reacts With HCl To Form Hydrogen And Magnesium Chloride. Write The Balanced Net Ionic Equation For This Reaction. 3. 4th, 2024

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(b) This Is A Chemical Reaction As A Gas Is Created (carbon Dioxide) During The Reaction. (a) + Hacog (b) The Bottle Could Be Heated Up Or Opened To Reverse The Reaction. CHECK YOUR LEARNING Suggested Answers 1. The Arrow Shows Which Way The Chemical Reaction Is Going. 2. 1th, 2024

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