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2 20) Acidic; 0.0758 M K<sub>3</sub>PO<sub>4</sub> 5) 20.8 ML AgNO<sub>3</sub> 21) 247 ML KOH 6) 166 ML AlCl<sub>3</sub> 22) 0.0556 M BaCl<sub>2</sub> 7) 13th, 2024

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8 Example 4.8: What Volume Of 0.150 M KCl Is Required To Completely React With 0.150 L Of 0.175 M  $\text{Pb}(\text{NO}_3)_2$  In The Reaction  $2 \text{KCl}(\text{aq}) + \text{Pb}(\text{NO}_3)_2(\text{aq}) \rightarrow \text{PbCl}_2(\text{s}) + 2 \text{KNO}_3(\text{aq})$ ? 0.1 10th, 2024

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