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VandenBout/LaBrake Noble Gas Notation: [Ar] 4s¹ 3d¹⁰

B) Chromium Orbital Notation: 1s² 2s² 2p⁶ 3s² 3p⁴ 4s¹ 3d⁵ Orbital Notation + Arrows: Noble Gas

Notation: 1[Ar] 4s 3d⁵ C) Molybdenum Orbital Notation: 1s² 2s² Mar 11th, 2024 Electron Configuration

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 Name The Elements Whose Electron Configurations
 Are: Element 2a) $61s^2 22s 62p 3s 3p$ 6b) $21s^2 62s^2$
 $22p 3s 3p 4s 3d^3$ 4. Name The Elements Whose Box
 Electron Configurations Are: Element 5. Without
 Looking At The Periodic Table, Write The Spdf Notation
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 Following Atomic Numbers: A) 10 B) 22 Mar 9th,
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 Configuration Of Elements. Note The 3d Orbital Follows
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 It Is Next To The 3p Orbital. May 5th, 2024.
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 Configurations - Solutions Note: The Electron
 Configurations In This Worksheet Assume That
 Lanthanum (La) Is The First Element In The 4f Block
 And That Actinium (Ac) Is The First Element In The 5f
 Block. If Your Periodic Table Doesn't Agree With This,
 Your Answers For Elements Near The F-orbitals May Be
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 Atomic Number # Of E-electron Configuration F F1- Or

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