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Reaction With O 2: C 2 H 5 OH (g) + 3O 2(g)  $\rightarrow$  2CO 2(g) + May 10th, 2024

# Molar Enthalpy Worksheet - Side A Q=mΔH

(Q= Energy Or Heat In KJ, M=moles,  $\Delta H$ =molar Enthalpy ) A Change In Enthalpy ( $\Delta H$ ) Is A Measurement Of E Nergy Transfer In The Form Of Heat . M Olar Enthalpy Is The Enthalpy Change Per Mole Of A Substance Involved In A Transformation. Examples Of Transformations Are Phase Changes, Dissolving, And Chemical Reactions. Apr 1th, 2024

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3. (1 Mark Each) When 10.0 G Of Ammonium Nitrate Is Dissolved In 100 Cm3 Of Water, The Temperature Of The Solution Decreases From 19.0°C To 10.5°C. A

Compare The Total Enthalpy Of The Reactants With That Of The Products And State Which Is Greate Mar 14th, 2024

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