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The Electrolysis Of Aqueous Silver Nitrate ( $\text{AgNO}_3$ ), For Example, Produces Oxygen At The Anode And Silver Metal At The Cathode. The Products Of The Reaction Demonstrate That Reduction Of Silver Ions ( $\text{Ag}^+$ ) To Silver ( $\text{Ag}$ ) Occurs More Readily Than Reduction Of Water. The Overall ...  $2\text{Ag}^+ + 2\text{I}^- \rightarrow 2\text{Ag}(s) + \text{I}_2(aq)$  ... 4th, 2024

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