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1. Examples of Free Energy Calculations Free Energy Calculations Are Carried Out In Much The Same As Enthalpy Calculations. There Are Tables Of Standard Free Energies Of Formation Of Compounds. Elements In Their Standard States Are Assigned Zero As Their Gibbs Free Energy Of Formation. Consider The Following Chemical Reaction: H₂(g) + Cl₂(g) → 2HCl(g). May 15th, 2024

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1 Chemical Reactions 1.1 Rate Of Reaction And Dependence On Temperature We Will Once Again Look At The Formation Of Ammonia (NH₃) From Nitrogen And Hydrogen (see Section Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The Equation: $N_2 + 3H_2 \rightarrow 2NH_3$ (1) $H_0 = 92 \text{ KJ Mol}$ $S_0 = 192 \text{ J Mol K}$ Feb 17th, 2024

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