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Introduction To Combustion 87 The Total Number Of Moles, N , Is The Sum Of The Number Of Moles Of All The Components Of The Mixture: $N = N_1 + N_2 + \dots$ It Follows That The Sum Of All The Mole Fractions Of The Mixture Must Also Equal 1. $X_1 + X_2 + \dots = 1$ The Mass Of Component i In A Mixture Is The Product Of The Number Of Moles Of i And Its Molecular Weight, M_i . The Mass Of The Mixture Is Therefore The Sum, $M = N_1M_1 + N_2M_2 + \dots$ 14th, 2024
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Application Of Gas Chromatography In Food Analysis Tical Chemistry Of Food Analysis [1], And Parti-cular Food Applications Involving GC, Such As Carbohydrates And Amino Acids [2], Lipids And Accompanying Lipophilic

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Grateful To Per Johansson, Erik Mellander, Harald Niklasson And Seminar
Participants At IFAU And IUI For Helpful Comments. Financial Support From The

Institute Of Labour Market Policy Evaluation (IFAU) And Marianne And Marcus Wallenbergs Stiftelse Is Gratefully Acknowledged. * Corresponding Author. IUI, Box 5501, SE-114 85 ... 15th, 2024.

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