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Agent And Reducing AgentWhen (b) Earn Electron, It Is Reduced, And It Is Therefore
An Oxidizing Agent. (A) Is Oxidized And (b) Is Reduced. The Answer Is C: In A Redox
Reaction, There Is Always An Oxidizing And Reducing Agent (no_3 ^ -) It Is Very
Likely To Be A Strong Oxidizing Agent. (NH_3) Is More Likely To Be Mar 6th, 2024Air
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