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0.125 Mole Apr 9th, 2024

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$10 \text{H}_2\text{O}$, Show What The Following Molar Ratios Should

Be. A. $\text{C}_4\text{H}_{10} / \text{O}_2$ B. O_2 / CO_2 C. $\text{O}_2 / \text{H}_2\text{O}$ D. $\text{C}_4\text{H}_{10} / \text{CO}_2$

E. $\text{C}_4\text{H}_{10} / \text{H}_2\text{O}$ 2. Given The Following

Equation: $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$ A. How Many

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