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 K F H ° KJ/mol F G ° KJ/mol S ° J/mol K C P J/mol K  
 Substances Not Containing Carbon: Ac Actinium 0.0  
 56.5 27.2 406.0 366.0 188.1 20.8 Ag Silver 0.0 42.6 ...  
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1. Examples of Free Energy Calculations Free Energy Calculations Are Carried Out In Much The Same As Enthalpy calculations. There Are Tables Of Standard Free Energies Of Formation Of Compounds. Elements In Their Standard States Are Assigned Zero as their Gibbs Free Energy Of Formation. Consider The Following Chemical Reaction:  $\text{H}_2(\text{g}) + \text{Cl}_2(\text{g}) \rightarrow 2\text{HCl}(\text{g})$ . 11th, 2024

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Standard Enthalpy Of Formation And Energy Of HOMO. The Second PC Is Strongly Influenced By Energy Of LUMO, While Substitution Pattern Parameters, Number Of Chlorine Atoms At  $\alpha$ -positions And Vicinal (adjacent) Carbon Atoms 16th, 2024

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Other than The standard state, Value of  $\Delta H_f^\circ$  represents the apparent Enthalpy Of the

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Example 9-4. Absorption Of SO<sub>2</sub> The Molar Density Of  
The Feed Gas May Be Determined From The Formula  
(mol/L)=12.2P(atm)T-1(K) Where P And T Are The  
Absolute Pressure And Temperature Of The Gas. The  
Following Data Are Taken For The Feed Gas: T=750F,



$P=150$  Psig,  $H=210$  Mm,  $R=82.4$  And For The Outlet Gas,  $R=11.6$ . Determine The (a) Mass Fraction Of SO<sub>2</sub>  
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Ash Drum, The Size Of The Drum And The Liquid And Vapor Compositions And Ow Rates. Which Of These Variables Are Speci Ed And Which Need To Be Chosen Depend On The Application<sup>2</sup>. Normally, The Feed Is Speci Ed, Ie. The Ow Rate  $F$ , Composition  $Z$ , Pressure  $P$   $F$  And Enthalpy  $H$   $F$  Are Known 16th, 2024

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