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 $2 \rightarrow$ CaSiO 3 B. 3Fe 2 O 3 + CO \rightarrow 2Fe 3 O 4 + CO 2 C. FeO +SiO 2 \rightarrow FeSiO 3 D. FeO \rightarrow Fe + 1 2 O 2 A. A B. D

C. C And D D. A And D Answer: C Solution: In Metallurgy Of Iron, CaO Is Used As Flux Which Is Used To Remove The Impurities Of SiO 2, CaO + SiO 2→ CaSiO 3. Also Here Fe 2 O 3 Is Reduced By CO To Fe 3 O 4 Which Is Further Reduced To ... Jan 1th, 2024

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