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 $2 \rightarrow \text{CaSiO}_3$
 B. $3\text{Fe}_2\text{O}_3 + \text{CO} \rightarrow 2\text{Fe}_3\text{O}_4 + \text{CO}_2$
 C. $\text{FeO} + \text{SiO}_2 \rightarrow \text{FeSiO}_3$
 D. $\text{FeO} \rightarrow \text{Fe} + \frac{1}{2}\text{O}_2$
 A. A B. D C. C And D D. A And D
 Answer: C
 Solution: In Metallurgy Of Iron, CaO Is Used As Flux Which Is Used To Remove The Impurities Of SiO₂,
 $\text{CaO} + \text{SiO}_2 \rightarrow \text{CaSiO}_3$. Also Here Fe₂O₃ Is Reduced By CO To Fe₃O₄ Which Is Further Reduced To ...
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