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A) [Co(H₂O)₆]²⁺ Co + 2 E) [FeO₄]²⁻ Fe + 6 I) K₂[CoCl₄] Co + 2 B) [CrCl₆]³⁻ Cr + 3 F) [Mn(CN)₆]⁴⁻ Mn + 2 J) K₃[AuF₆] Au + 3 C) [Co(NH₃)₆]Cl₂ Co + 2 G) [Ni(CO)₄] Ni 0 K) (NH₄)₂[IrCl₆] Ir + 4 D) [Co(NH₃)₅]Cl₂ Co + 3 H) [Ni(EDTA)]²⁻ Ni + 2 L) Na[Mn(CO)₅] Mn -1 TASK 2 - Writing Half Equations A) 2I⁻ → I₂ + 2e⁻ B ... 3th, 2024

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Metal Carbonate → Metal Oxide + Carbon Dioxide (on Heating) CaCO₃ → CaO + CO₂ TASK 4 - WRITING BALANCED EQUATIONS 1) Balance The Following Equations. 11th, 2024

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(A2 BBoF 23) 4 Kc For The Equilibrium Below Is 7.50 At 50 °C. If 50.0 G Of Ethanol Is Mixed With 50.0 G Of Propanoic Acid, What Mass Of Ethyl Propanoate Will Be Formed At Equilibrium? CH₃CH₂OH(l) + CH₃CH₂COOH(l) ⇌ CH₃CH₂COOCH₂CH₃(l) + H₂O(l) ((((((6th, 2024

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