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1. Solubility Of Ethanol In Water  
Explain Why Ethanol ( $\text{CH}_3\text{CH}_2\text{OH}$ ) Is More Soluble In Water Than Is Ethane ( $\text{CH}_3\text{CH}_3$ ).  
Answer Ethanol Is Polar; Ethane Is Not. The Ethanol  $-\text{OH}$  Group Can Hydrogen-bond With Water.  
2. Calculation Of PH From Hydrogen Ion Concentration  
What Is The PH Of A Solution That Has An  $\text{H}^+$  Concentration Of (a)  $1.75 \times 10^{-5}$  Mol/L; (b)  $6.50 \times 10^{-10}$  Mol/L; (c)  $1.0 \times 10^{-4}$  Mol/L;  
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