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1. Consider The Galvanic Reaction Between The Two Half Cells, $\text{Pb}|\text{Pb}^{2+}(\text{aq})$ And $\text{Ag}|\text{Ag}^{+}(\text{aq})$.
1(a) (6 Pts) Determine The Spontaneous Reaction. Write A Balanced Overall Equation And Determine E° .
1(b) (4 Pts) Label The Electrodes And Solutions With The Following Labels: "Pb," "Pb²⁺," "Ag," And "Ag⁺." Make
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