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Between Positively Charged Ions (cations) And Negatively ... Covalent/metallic/ionic  
> H-bonds > Permanent-permanent Dipole > Instantaneous-induced Dipole ...

Answers: Q.1 Across Period 3, The Boiling Point (b.p.) Increases From Sodium To Aluminium, Since The ... 19th, 2024  
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HOW TO NAME COVALENT COMPOUNDS? NAME THE FOLLOWING COVALENT COMPOUNDS - ANSWERS Name Chemical Formula 1. Sulfur Dioxide  $\text{SO}_2$  2. Silicon Dioxide  $\text{SiO}_2$  3. Sulfur Trioxide  $\text{SO}_3$  4. Sulfur Dichloride  $\text{SCl}_2$  5. Nitrogen Trifluoride  $\text{NF}_3$  6. Dinitrogen Trioxide  $\text{N}_2\text{O}_3$  7. Dinitrogen Pentoxide  $\text{N}_2\text{O}_5$  8. Dichlorine Heptoxide  $\text{Cl}_2\text{O}_7$  9. Carbon 11th, 2024  
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Worksheet 4.2 C Naming Ionic And Covalent Compounds ... Topic 4: Chemical Bonding And Structure 4.1 Idea And Essential Structure: Ionic Compounds Are Made Up Of Ions Held Together In Reticulatory Structures By Ionic Bonds. Understanding: Positive Ions (cations) Form From Metals That Lose Valence Electrons. Negative ... 10th, 2024

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