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Lecture 3: Solubility Of Gases, Liquids, And Solids In Liquids

Case 3: Gases Mixing With Liquids Consider Specifically Gases In H₂O. We Know Gases Dissolve Because We Hear The Fizz From CO₂ And We Know That Fish Can Breathe O₂. But H₂O Is Polar, With Hydrogen Bonding And According To "like Dissolves Like" It Should Prefer ...File Size: 640KBPage Count: 8 May 9th, 2024

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- Intermolecular Forces Are Much Weaker Than Ionic Or Covalent Bonds (e.g., 16 kJ/mol Versus 431 kJ/mol For HCl).
- Melting O May 8th, 2024

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And ...

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Types Of Intermolecular Forces: Dispersion, Dipole-
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