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Directions: Solve Each Of The Following Problems.

Show Your Work, Including Proper Units, To Earn Full

Credit. 1. Silver And Nitric Acid React According To The

Following Balanced Equation: $3 \text{Ag}(s) + 4 \text{HNO}_3(aq) \rightarrow 3$

$\text{AgNO}_3(aq) + 2 \text{H}_2\text{O}(l) + \text{NO} \dots$ May 2th, 2024.

Chemistry Supplemental Stoichiometry Problems

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Problems Answer Key Stoichiometry Is A Chemical

Section Involving The Use Of Relationships Between

Reagents And / Or Products In A Chemical Reaction To

Determine The Desired Quantitative Data. In Greek,

Stoikhein Means Element And Metricity Means

Measurement, Then Steichiometry Literally Translated

Means ... Apr 1th, 2024Chemistry Stoichiometry

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Chemistry 11 Stoichiometry Review Worksheet Name: KEY
Chemistry 11 Stoichiometry Review Worksheet Name: KEY 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1 \text{ mol} = 122.55 \text{ g}$
 $\times 2 \text{ O}_2 = 1 \text{ KClO}_4 \times 32.00 \text{ g/mol} = 3.11 \text{ g O}_2$
% Yield = $\frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$ 34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percent purity = 62.8%? pure FeCO_3 Feb 4th, 2024
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General Information You Will Need A Periodic Table, Your Stoichiometry Notes,
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Properties Of The Elements: Chapter 8 Basic Concepts

Of Chemical Bonding: Chapter 9 Molecular Geometries

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