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Mole Fraction Water Of Solution $\chi_{\text{Water}} = (55.6 \text{ Mol}) / (0.292 + 55.6) = 0.995$ (or 1-

χ Sugar) Molality $M = (.292 \text{ Mol})/(1 \text{ Kg}) = 0.292 \text{ Mol Kg}^{-1}$ Molarity $M = (.292 \text{ Mol})/(1 \text{ Mar 1th, 2024.}$

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