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Naming Acids 1. NAMING BINARY ACIDS NAMING OXYACIDS

1. NAMING BINARY ACIDS: The Name Of The Binary Acid Consists Of Two Words. The First Word Has Three Parts: The “hydro” Prefix The Root Of The Nonmetal Element The “ic” Ending The Second Word Is Always “acid” Examples: HCl = Hydro Chloric Acid = Hydrochloric Acid HBr = Hydro Bromic Acid = Hydrobromic Acid 3th, 2024

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Worksheet 6 - Naming Ionic Compounds Part 2 (Binary Compounds With Group 3 - 12 Metals) Naming (STOCK SYSTEM) ... AuP 8. Au 3 P 9. FeO 10. Sn 3 P 4 11. Fe 2 C 12. Co 4 C 3 13. Hg 3 N 14. MnS 2 15. AsI 5 16. SnO 17. Ti 2 S 3th, 2024

Naming Chemical Compounds Part One: Naming Ionic ...

- Is NOT An Ionic Compound; It Does Contain A Metal (Mn) And A Non-metal (O), But It Has A Charge. Thus, It Is A Polyatomic Ion, Not A Compound. A Compound Will NEVER Have A Charge! Naming Ionic Compounds There Are Three Steps Involved In Naming Ionic Compounds- Naming The Cation, Naming The Anion, And 2th, 2024

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Handout: Naming Organic Compounds

Prefix + Parent + Suffix Group 3 CH 1. Name Parent+suffix: Longest Carbon Chain + Family Suffix. 2. Number Carbons In Parent Chain: Begin Numbering From End That Meets Specified Criteria (*See Nomenclature Chart). 3. Name Prefix: Substituent Position #s And Names (grouping 3th, 2024

Naming Compounds Handout

CHEMISTRY Naming Compounds Handout Page 6 Of 12 CHEMICAL FORMULAS AND NAMES FROM INDIVIDUAL IONS Compounds Are Named From The Individual Ions They Come From. Name The Cation And The Anion, Then Remove “ion” From Each Name: E.g. Na + = Sodium Ion Cl - = Chloride Ion \Rightarrow NaCl = Sodium Chloride K + = Potassium Ion CO₃ ⁻² 3th, 2024

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An Ion Formed From An Individual Atom Is A Monatomic (or Monoatomic) Cation. I. Groups IA To IIA Elements, Aluminum (Al), Silver (Ag), Zinc (Zn) And Cadm 2th, 2024

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Answers - Naming Chemical Compounds

Answers - Naming Chemical Compounds . Name The Following Chemical Compounds: 1) NaBr Sodium Bromide. 2) $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$ Calcium Acetate. 3) P_2O_5 Diphosphorus Pentoxide. 4) $\text{Ti}(\text{SO}_4)_2$ Titanium(IV) Sulfate. 5) FePO_4 Iron(III) Phosphate. 6) K_3N Potassium Nitride. 7) S 1th, 2024

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