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# Statistics Of Nuclear Decay "Half-life Decay Formula"

The Number Of Recorded Counts. This Variation Is Called The Statistical Uncertainty Of A Measurement. It Is An Uncertainty That Is Well Understood. Consider A Measurement That Is Done For A Time Period  $\Delta t$ . Suppose That There Are N Radioactive Nuclei, And That Each One Has A Probability P To Decay In The Time May 7th, 2024

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Think Of As Hydrogen, Is Called Hydrogen-1 And Is Stable. The Other Two Isotopes Are Called Hydrogen-2 (deuterium) And Hydrogen-3 (tritium). Deuterium Apr 21th, 2024

## **Introduction To Nuclear Physics And Nuclear Decay**

The Binding Energy Per Nucleon Dictates Results When Nuclides Break Apart (fission) Or Fuse Together (fusion) (keep In Mind That ... (two Protons And Two Neutrons) ! "=  $2 \text{ 4He} + 2 \dots$  Or An Anti-neutrino () Neutrinos Have No Charge, Spin 1/2, And Mass  $\sim 0.1 - 1 \text{ EV}$  (?) The Beta Particle Apr 15th, 2024

#### **CHAPTER 1 Radioactive Decay & Decay Modes**

This Equation Represents A First-order Reaction And In The Present Case A Mononuclear Reaction. Radioactive Decay Is Only Possible If  $\Delta E > 0$ .  $\Delta E$  Can Be Determined By Comparison Of The Masses. According To The Relation By Einstein (E = Mc2): (EQ 3) By Calculation Of  $\Delta E$  It C Jan 13th, 2024

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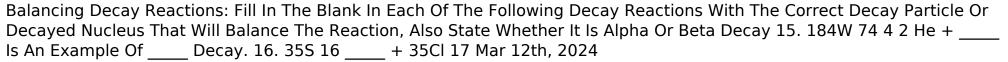
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