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(c) Write The Ionic Equation For The Reaction In Which Lead(II) Iodide Is Formed..... [2] (d) Aqueous Silver Nitrate Produces A Yellow Precipitate With Both Iodide Ions And Carbonate Ions. When Testing An Unknown Solution For Iodide Ions, The Aqueous Silver Nitrate Is Acidified. Explain Why The Aqueous Silver Nitrate Is ... Jan 9th, 2024

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(d) Barium Chloride Is An Ionic Compound. Draw A Diagram That Shows The Formula Of The Compound, The Charges On The Ions And Gives The Arrangement Of The Valency Electrons Around The Negative Ion. The Electron Distribution Of A Barium Atom Is 2.8.18. Mar 17th, 2024

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