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Concentration Of 0.300 M? A) 0.161 B) 0.260 C) 0.200

D) 0.225 E) 0.245 16. What Is The Half Life In Sec For A

First Order Reaction Wiht  $K = 2.03 \times 10^{-3} \text{ S}^{-1}$ ? A) 323

B) 369 C) 341 D) 422 E) 395 17. The Half Life Of A First

Order Reaction Is 45.6 Sec. Wha Jan 17th,

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Physical ...Agarose Physical Chemistry Appendix B:

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