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Will remain after 74.4 sec for an initial concentration of 0.300 M? A) 0.161 B) 0.260 C) 0.200 D) 0.225 E) 0.245
16. What is the half-life in sec for a first-order reaction with $k = 2.03 \times 10^{-3} \text{ s}^{-1}$? A) 323 B) 369 C) 341 D) 422 E) 395
17. The half-life of a first-order reaction is 45.6 sec. What Apr 22th, 2024

Appendix B: Agarose Physical Chemistry Agarose Physical ...

Agarose Physical Chemistry Appendix B: Agarose Physical Chemistry continued. The energy needed to melt an agarose gel increases as the gel concentration increases. The gelling temperature of an agarose gel is also influenced by the Mar 28th, 2024

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