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State The Oxidation Number Of Each Of The Elements That Is Underlined. A)  $\text{NH}_3$  -3 B)  $\text{H}_2\text{SO}_4$  6 C)  $\text{ZnCO}_3$  4 D)  $\text{Al}(\text{OH})_3$  3 E)  $\text{Na}_2\text{O}$  0 F)  $\text{Cl}_2$  0 17. Balance The Redox Equation Using The Half Reaction Method.  $\text{Al} + 3\text{Ag}^+ \rightarrow \text{Al}^{3+} + 3\text{Ag}$  18. Ci May 3th, 2024

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