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Design, Is Available In Single Or Dual Pass. 18th, 2024Chapter 20 • Heat • Heat Transfer • Phase Changes Specific ... The Specific Heat Of A Substance Is The Amount Of Energy It Requires To Raise The Temperature Of 1 Kg, 1 Degree Celsius. Q Mc T= Δ 0 QI C M T Kg C = = Δ · •The Higher The Specific Heat, The More Energy It Takes And The Longer It Takes To Heat Up And To Cool Off. • The Lower The Specific Heat, The Less Energy It Takes And The 23th, 2024Heat Transfer And Heat Capacity - Virginia1. Pass Out The Attached "Heating Curve" Lab Worksheet, And Let Students Read It Over. Go Over The Procedures And Safety Issues Involved In The Lab: Hot Plates Can Become Very Hot, So Be Careful Not To Touch Them. Exercise Care With Hot Water And Beakers; Use Wire Mesh To Set Beake 3th. 2024.

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