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Interacts With Solvent Molecules. The Tert-butoxide Ion Is Very Bulky, Or Sterically Hindered, And Is Less Capable Of Interacting With The Solvent. The Ethoxide Ion Is Not As Sterically Hindered So It Can Accommodate More Solvent Interactions. As A Result Ethoxide Is Better Solvated And Is Therefore More Stable Than Tert-butoxide (Figure 3.7). May 25th, 2024.

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Same Two Chemicals In Two Different Ways. The Formula That Shows The Bonds As Lines And What Atom Is Connect To What Are Called "structural Formulas Because From The Formula You Can See How The Atom's Are Attached To Each Other. The Lower Formulas Are Called Condensed Formulas . Apr 3th, 2024.

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