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Determining Molarity Of A Solution

Determining An Equilibrium Constant Using Spectrophotometry And Beer's Law
Objectives: 1.) To Determine The Equilibrium Constant For The Reaction Of Iron (III)
And Thiocyanate To ... The Molarity Of The Diluted KSCN Solution In The 250 ml
Beaker Before Any Cu^{2+} Was Added: $(10.0 \text{ ml}) (5.21 \times 10^{-2} \text{ Mmole/ml}) = 5.21 \times 10^{-1} \text{ Mmole/ml}$
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