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Determining An Equilibrium Constant Using Spectrophotometry And Beer's Law Objectives: 1.) To Determine The Equilibrium Constant For The Reaction Of Iron (III) And Thiocyanate To ... The Molarity Of The Diluted KSCN Solution In The 250 MI

Beaker Before Any Cu2+ Was Added: (10.0 MI) (5.21 X 10-2 Mmole/ml) =  $5.21 \times 10-1 \text{ Mmole/ml}$  Jan 1th, 2024

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