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25. What Is The Empirical Formula Of A Compound That
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A) SO B) SO₂ C) SO₃ D) SO₄ 26. A Compound Consists
Of 40.% Sulfur And 60.% Oxygen 17th, 2024 Chemistry
11 Stoichiometry Review Worksheet Name:

KEY Chemistry 11 Stoichiometry Review Worksheet
Name: ___KEY___ 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1 \text{ mol}$
 $122.55 \text{ g} \times \frac{1 \text{ mol}}{122.55 \text{ g}} = 0.00816 \text{ mol}$
 $0.00816 \text{ mol} \times 2 \times 32.00 \text{ g/mol} = 0.528 \text{ g O}_2$
% Yield = $\frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$ 34. $4\text{FeCO}_3 + \text{O}_2$
 $\rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percent purity = 62.8%? g pure FeCO_3 23th,
2024 Chemistry 12, Stoichiometry Review Problems, R.
Janssen ... In A Camping Stove? 2. How Many Moles Of
 O_2 Molecules Should Be Supplied To Burn 1 Mol Of CH_4
... Was At One Time Considered For Use As A Rocket
Fuel. ... How Many Formula Units Of Calcium Oxide
(CaO) 10th, 2024.

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