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Titration Of Strong Acid With Strong Base
0 2 4 6 8 10 12 14 0 10 20 30 40 50 MI Base Added
PH Eq. Pt. PH=7.00 F A=0.01 M V A=40 MI F B=0.02 M At Equiv. Point: V B=(V A*M A)/M B=20 MI Strong Acid / Strong Base Titration Remaining Strong Acid Determines PH Excess Strong Base Determines POH And, Hence, The PH Formal Amount Of Acid In Initial Solution Apr 8th, 2024

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