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Thermochemical Equations - Sayre Chemistry  $Q = mc\Delta T$  Where M Is The Mass Of The Substance C Is The Object's Specific Heat And T Is The Change In Temperature (final - Initial) Parts 2 And 4 Cannot Be Calculated With This Equation, Because There Is No Change In Temperature. Instead, They Need To Be Calculated Using Their Latent Heats Of Vaporization/fusion 7th, 2024 AP Chemistry Practice Test, Ch. 6: Thermochemistry ... AP Chemistry Practice Test, Ch. 6: Thermochemistry Name \_\_\_\_\_ MULTIPLE CHOICE. Choose The One Alternative That Best Completes The Statement Or Answers The Question. 1) A Chemical Reaction That Absorbs Heat From The Surroundings Is Said To Be \_\_\_\_\_ And Has A \_\_\_\_\_  $\Delta H$  At Constant Pressure. A) endothermic, Positive 2th, 2024 Chapter 10 Practice Worksheet: Thermochemistry: Chemical ... Name: \_\_\_\_\_ Section: \_\_\_\_\_ Chapter 8 Worksheet Spring 2007 Page 1 Of 4 Chapter 10 Practice Worksheet: Thermochemistry: Chemical Energy 1) Describe The Difference Between Potential Energy And Kinetic Energy. 2) What Is The Difference Between Heat And Temperature 13th, 2024.

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