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(d) The Oxidation States Of Vanadium In Its Compounds Are V(+5), V(+4), V(+3) And V(+2). The Vanadium(III) Ion Can Behave As A Reductant Or An Oxidant. (i) Indicate On The Following Equation Which Reactant Is The Oxidant. $2V^{3+} + Zn \rightarrow 2V^{2+} + Zn^{2+}$ [1] (ii) Which Change In The Following Mar 13th, 2024

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(d) Molten Zinc Chloride Can Be Electrolysed Using The Apparatus Shown Below. +-
A C D B (i) Which One Of The Letters, A, B, C Or D, Represents The Cathode?..... [1]
(ii) Which One Of The Following Substances Is The Most Suitable For Use As An
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(iii) Copper Can Be Extracted By The Electrolysis Of An Aqueous Solution. Suggest
Why The Electrolysis Of An Aqueous Solution Cannot Be Used To Extract
Aluminium..... [2] 2-(b) The Ions Which Are Involved In The Electrolysis Are Al^{3+}
And O^{2-} . The Products Of This Electrolysis Are Given On ... Mar 24th, 2024

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Explain How Aqueous Sodium Hydroxide Can Be Used To Separate ... Alkenes Can
Be Oxidised To Carboxylic Acids By Boiling With Aqueous Potassium Manganate(VII).
(i) Pent-2-ene, $CH_3-CH_2-CH=CH-CH_3$, Oxidises To CH_3-CH_2-COOH And CH_3
... 79 Se Selenium 34 80 Br Bromine 35 84 Kr Krypton 36 39 K Potassium 19 88 Sr

Strontium 38 89 Y ... Apr 2th, 2024

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Any Correct Equation Producing Two Alkenes And Hydrogen, E.g. $C_2H_4 + C_5H_{10} + H_2$ / $C_3H_6 + C_4H_8 + H_2$; 1 4(b)(iv) Alkenes: More Useful Than Alkanes / Used To Make Polymers Or Plastics / Used To Make Chemicals / Petrochemicals; Or Alkanes: (balance The Demand For Different) Fue Feb 24th, 2024

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(c) Write The Ionic Equation For The Reaction In Which Lead(II) Iodide Is Formed.....
[2] (d) Aqueous Silver Nitrate Produces A Yellow Precipitate With Both Iodide Ions
And Carbonate Ions. When Testing An Unknown Solution For Iodide Ions, The
Aqueous Silver Nitrate Is Acidified. Explain Why The Aqueous Silver Nitrate Is ... Feb
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(c) Write The Ionic Equation For The Reaction In Which Lead(II) Iodide Is Formed.....
[2] (d) Aqueous Silver Nitrate Produces A Yellow Precipitate With Both Iodide Ions
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